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New Lipophilic Derivatives of AZT and d4T 5'-Phosphonates

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New Lipophilic Derivatives of AZT and d4T 5'-Phosphonates

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ABSTRACT

5'-Aminocarbonylphosphonyl and aminocarbonylmethylphosphonyl diesters of AZT and d4T were synthesized as potential anti-HIV agents.

Key Words: Anti-HIV agents; AZT and d4T derivatives; Nucleoside 5'-phosphonates.

Going on the studies of antiviral agents in the series of nucleotide derivatives, we paid attention on AZT conjugated with phosphonoformic (PFA) and phosphonoacetic (PAA) acids. Earlier diesters and triesters of type (I) were prepared. [1–3] Their antiviral effect was much higher than that of PFA and comparable with that of AZT. A poor increase in the efficacy could be due to high instability in water solutions of the triesters and poor penetration through cell membranes of the diesters. [3]

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We synthesized a set of uncharged 5'-aminocarbonyl- and 5'-aminocarbonyl-methylphosphonate derivatives of AZT and d4T (II-V) and evaluated their activity in cell cultures infected with HIV-1.

AZT-derived				d4T-derived			
Comp	n	R'	R	Comp	n	R'	R
IIa IIb IIc	0 0 0	H Me PhCH ₂ CH ₂	Et Et Et	IVa IVb IVc	0 0 0	H Me PhCH ₂ CH ₂	Et Et Et
IIIa IIIb IIIc	1 1 1	H H Me	Et cyclohexyl Et	Va _ _	1	Н	Et

The target amides **II–V** were prepared by treatment of the corresponding 5′-ethoxycarbonylphosphonyl nucleosides with primary amines followed by esterification with various alcohols. Esterification was carried out using two procedures: in the presence of TPSCl and under Mitsunobu conditions. In the case of secondary alcohols, the first method was preferential, whereas for primary alcohols the second procedure resulted in higher yields.

The resulting compounds were mixtures of diastereoisomers (due to a chiral phosphorus atom). Isomers A and B of amide **Ha** were separated by reverse-phase chromatography, their retention times differing in about 2 min.

Chemical stability of the synthesized amides was estimated in 0.05 M phosphate buffer at 37°C. Figure 1 shows the stability of amide **Ha** under various pH. Obviously, it was strongly pH-dependent. In the range of pH 5.5–7.8 its half-life time decreased from 6 h to less than 30 min. At physiological pH 7.2–7.4 its half-life time was about 1 h.

The composition of the reaction mixture in the process of chemical hydrolysis of amide **Ha** was studied at pH 7.2 at 37°C. We observed the formation of two products, a major one being AZT and a minor one - 5′-aminocarbonylphosphonyl-AZT. The formation rate of the former was considerably higher than that of the latter (Fig. 2). The corresponding PAA-AZT analogue was degraded under similar conditions to give only the nucleoside.

As a whole, PAA derivatives in both series were noticeably more stable than the corresponding PFA ones (for example, 7 h for IVa vs. \gg 24 h for V). d4T derivatives

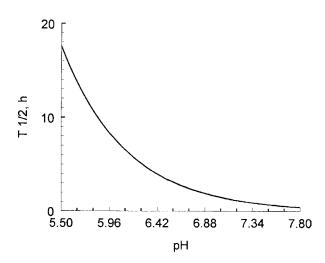


Figure 1. pH-Dependence of half-life time of amide IIa.

had longer half-lives than the corresponding AZT derivatives (7 h for **IVa** against 2.5 h for **IIa**). The substitution at the amide group increased the chemical stability (2.5 h for **IIa**, 6 h for **IIb** and > 10 h for **IIc**).

Antiviral effect of the compounds under study was evaluated in MT-4 cells infected with HIV-1 strain GKV-4046 (Table 1). The tested compounds were added into the culture after the virus adsorption. The activity was measured by the amount of viral p24 antigen using immunoassay. Cytotoxicity was measured by the colorimetric method.

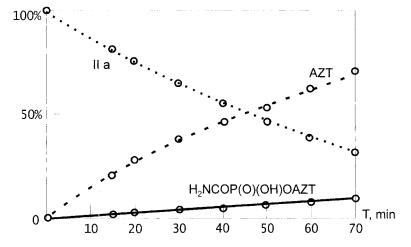


Figure 2. Chemical hydrolysis of amide IIa in phosphate buffer at pH 7.2.

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Table 1. Anti-HIV properties of the synthesized compounds.

		CD ₅₀ * (μM)	ID ₅₀ ** (μM)	ID ₉₀ ** (μM)	IS***
	$ \begin{array}{c} \mathbf{IIa} \\ \mathbf{R} = \mathbf{Et}, \ \mathbf{R}^1 = \mathbf{H} \end{array} $	298.3	0.0009	0.004	346800
$\begin{matrix} O & O \\ R^{\dagger}NH-C-P-O \\ OR \end{matrix} \begin{matrix} Thy \\ IIa-c \end{matrix}$	IIb $R = Et, R^{1} = Me$	112.9	0.00024	0.007	470400
Ila-c N_3	IIc $R = Et,$ $R^{1} = PhCH_{2}CH_{2}$	177.7	< 0.0002	0.0002	> 888500
0 0	$IIIa$ $R = Et, R^1 = H$	456.3	0.29	1.68	1573
$\begin{array}{c} O & O \\ I \\ R^1NH-C-CH_2-P-O \\ IIIa,c \\ OR \\ N_3 \end{array} $	IIIc $R = \text{cyclohexyl},$ $R^{1} = H$	170.1	0.0085	-	20012
0 0	IVa R = Et, R1 = H	573.5	0.9	2.1	637
O O Thy	IVb $R = Et, R^{1} = Me$	85.7	0.04	0.27	2187
$\begin{array}{c} O & O \\ I & I \\ R^{1}NH-C-CH_{2}-P-O \\ V & OR \end{array}$	\mathbf{V} $\mathbf{R} = \mathbf{Et}, \ \mathbf{R}^1 = \mathbf{H}$	267.9	5.4	18.8	50
AZT d4T		187.5 314	0.018 0.24	0.35 23.8	11030 1308

^{*}CD₅₀ – Compound concentration required for 50% inhibition of cell proliferation.

As is seen from the table, this modification seems to be promising in the case of PFA-AZT series: as a rule a significant increase in the activity resulted in an increase of the selectivity index. For d4T analogues, this gain was not so obvious.

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^{**} \mathbf{ID}_{50} and \mathbf{ID}_{90} – Compound concentration required for 50% or 90% inhibition of HIV reproduction, respectively.

^{***}SI – selectivity index, CD₅₀/ID₅₀.

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